

# **Patent and Trademark Office**

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Washington, D.C. 20231

APPLICATION NO.	FILING DATE	FIRST NAMED	INVENTOR	·	ATTORNEY DOCKET NO.
08/634,255	04/18/96	OHKUMA		N ·	35.C11365
			· _	EXAMINER	
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FITZPATRICK CELLA HARPER & SCINTO				ANNICK, C	
30 ROCKEFELLER PLAZA				ART UNIT	PAPER NUMBER
NEW YORK NY	10112-380	01			<u> </u>
				2853	
				DATE MAILED:	
					01/04/99

Please find below and/or attached an Office communication concerning this application or proceeding.

Commissioner of Patents and Trademarks

## Office Action Summary

Application No. 08/634,255

Applicant(s)

Examiner

Christina Annick

Group Art Unit 2853

Ohkuma et al.



X Responsive to communication(s) filed on Oct 13, 1998	
This action is <b>FINAL</b> .	
Since this application is in condition for allowance except for f in accordance with the practice under Ex parte Quayle, 1935	C.D. 11; 453 O.G. 213.
A shortened statutory period for response to this action is set to is longer, from the mailing date of this communication. Failure to application to become abandoned. (35 U.S.C. § 133). Extension 37 CFR 1.136(a).	expire 3 month(s), or thirty days, whichever o respond within the period for response will cause the
Disposition of Claims	
X Claim(s) 1, 2, and 4-15	is/are pending in the application.
Of the above, claim(s)	is/are withdrawn from consideration.
☐ Claim(s)	is/are allowed.
X Claim(s) 1, 2, and 4-15	is/are rejected.
☐ Claim(s)	is/are objected to.
☐ Claims	are subject to restriction or election requirement.
Application Papers	
<ul> <li>See the attached Notice of Draftsperson's Patent Drawing</li> <li>☐ The drawing(s) filed on</li></ul>	ed to by the Examiner.  isapproveddisapproved.  under 35 U.S.C. § 119(a)-(d).  f the priority documents have been  mber)  International Bureau (PCT Rule 17.2(a)).
Attachment(s)  Notice of References Cited, PTO-892  Information Disclosure Statement(s), PTO-1449, Paper No.  Interview Summary, PTO-413  Notice of Draftsperson's Patent Drawing Review, PTO-94  Notice of Informal Patent Application, PTO-152	
SEE OFFICE ACTION ON T	THE FOLLOWING PAGES

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#### **DETAILED ACTION**

### Claim Rejections - 35 USC § 103

- 1. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
  - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 2. Claims 1, 2, and 4-15 are rejected under 35 U.S.C. 103(a) as being unpatentable over Ohkuma et al. (U.S. 5,478,606) in view of Field et al. (U.S. 3,852,222).

Ohkuma et al. discloses a liquid jet recording head which includes a member formed from a cured product of a resin composition comprising an epoxy and a photopolymerization initiator which acts to cure the epoxy (see column 5, lines 35-60). The resin composition which includes a compound which contains a functional group which reacts with the curable epoxy, as well as a fluorocarbon moiety, is subjected to cationic polymerization and (see column 5, lines 61-68 and column 6, lines 4-6). Note: Although the reference does not specifically disclose that this compound contains a reactive "functional group" this group is inherently disclosed since the reference does disclose that this compound "reacts" with the curable epoxy which in most cases involves a functional group (see column 6, lines 1-6). Tables 1 and 2 teach various concentrations of the compound having a functional group and fluorocarbon moiety. In addition,

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the compound which contains a fluorocarbon moiety (see column 6, line 6) contains fluorine at an amount of ~30% which is well within the range specified. A curing agent is also disclosed in column 5, lines 55-60. Further, the curable epoxy compound is an aromatic epoxy compound such as bisphenol A (see column 5, lines 35-36). The curable epoxy compound disclosed also includes an alicyclic epoxy compound which is an epoxy compound having an oxycyclohexane skeleton (see column 5, lines 35-42). The reference also discloses a method of making the liquid jet recording head which entails forming an ink flow path pattern form a soluble resin on an ink discharge pressure-generating element on a base plate, forming a coating resin layer on the soluble resin layer, removing of the soluble resin layer by elution, and forming a discharge opening through the coating resin layer (see column 2, lines 28-42). In addition, the reference discloses the a the method of forming the discharge opening is accomplished by the well known technique of photolithography (see column 4, lines 28-32). Finally, it is well known in the art to form discharge openings by either oxygen plasma etching or excimer laser irradiation. In any event, Ohkuma et al. discloses in column 7, lines 29-30 that the discharge openings can be formed by either of these methods.

Ohkuma et al. discloses the claimed invention except for the functional group which reacts with the curable epoxy being a hydroxyl group, where the compound having a functional group has the structure of an aromatic or alkyl fluorinated diol. Field et al. teaches that it is known to use aromatic and alkyl diols which contain hydroxyl functional groups as compounds useful for polymerization reactions as set forth in column 2, lines 13-20. It would have been obvious to a

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person of ordinary skill in the art at the time the invention was made to use aromatic and alkyl diols which contain hydroxyl functional groups as the compound which reacted with the curable epoxy of Ohkuma et al. as taught by Field et al. in order to provide a polymeric coating to the liquid jet recording head which is highly hydrophobic and therefore extends the life of the liquid jet recording head.

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Finally, although neither Ohkuma et al. or Field et al. teach the specific concentration (i.e., 5-50%) of the compound having a functional group and fluorocarbon moiety, both Ohkuma et al. (in Tables 1 and 2) and Field et al. (in Examples I and II) teach various concentrations of this compound. In addition, in a polymerization reaction involving functional groups such as hydroxyl groups, the weight of the compound having functional groups is irrelevant, since the reactivity of the compound having functional groups is based on the stoichiometric ratio of the hydroxyl groups on the molecule, so that a molecule that has many hydroxyl groups can be used at a lower weight percent than a molecule having few hydroxyl groups. Therefore, in the absence of any showing of criticality for this claimed range, it would have been obvious to one of ordinary skill in the art at the time the invention was made to utilize routine optimization when selecting the weight of the compound having a functional group of Ohkuma et al., for the purpose of optimizing the stoichiometric or molar ratios of that compound in the polymerization reaction.

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#### Response to Arguments

3. Applicant's arguments filed 10/13/98 have been fully considered but they are not persuasive.

Applicant urges that Ohkuma et al. does not disclose a compound having a functional group reactive to the epoxy compound and a fluorocarbon moiety. Applicant's attention is directed to Table 1 which is found in column 9 of Ohkuma et al., which discloses "butylphenoliodonium hexafluoroantimonate" as a "Cationic photopolymerization initiator". This compound constitutes a compound having a functional group and a fluorocarbon moiety reactive to the epoxy compound. Applicant urges that this photopolymerization initiator and copper triflate are not "fluorocarbons having a functional group reactive to the epoxy resin as defined in the present invention." However, Applicant does not substantiate this argument with any evidence as to why this assertion is correct.

Next, Applicant urges that the object of Field et al. is "to obtain high water repellency" and that this quality of the resin would cause a portion of an ink jet head "such as a flow path wall", to hold bubbles. However, Applicant is arguing limitations which are not in the claims. In addition, as previously stated, since the combination of Ohkuma et al. and Field et al. teaches the claimed resin composition and chemical structure of the compound having functional groups, it would be expected to perform with an ink jet head in a similar manner as the claimed resin composition.

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Conclusion

5. Any inquiry concerning this communication or earlier communications from the examiner

should be directed to Christina Annick whose telephone number is (703) 308-6398. The examiner

can normally be reached on Monday to Friday from 8:00 to 5:00.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Mr.

Benjamin Fuller, can be reached on (703) 308-1782. The fax phone number for this Group is

(703) 305-1341.

Any inquiry of a general nature or relating to the status of this application or proceeding should be

directed to the Group receptionist whose telephone number is (703) 308-1782.

۷ Christina Annick

C. anul

December 23, 1998

Benjamin R. Fuller Supervisory Patent Examiner Page 6

**Technology Center 2800**